## PRELIMINARY NOTE

Thermal Conversion of Fluorinated Azo-compounds into Indazoles : the Case of 2,5,6-Trifluoro-4-(2,4,6-trimethylphenylazo)pyrimidine

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## SUMMARY

Thermal dehydrofluorination of 2,5,6-trifluoro-4-(2,4,6-trimethylphenylazo)pyrimidine in boiling xylene to 2,4-difluoro-7,9-dimethyl-5<u>H</u>-pyrimido[4,5-<u>c</u>]benzo[1,2]diazepine is accompanied by the formation, <u>inter alia</u>, of 5,7-dimethyl-2-(2,5,6-trifluoropyrimidin-4-yl)-<u>2<u>H</u>-indazole. The same product is formed when tetrafluoropyrimidine is treated with 5,7-dimethylindazole.</u>

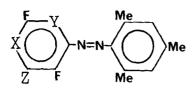
As pointed out previously [1], 2,5,6-trifluoro-4-(2,4,6trimethylphenylazo)pyrimidine (1) does not undergo clean thermal solution-phase intramolecular dehydrofluorination to 2,4-difluoro-7,9dimethyl-5<u>H</u>-pyrimido[4,5-<u>c</u>]benzo[1,2]diazepine (2), a situation which contrasts notably with the conversions (3)  $\rightarrow$  (4) (>80% yield\*\*; achieved in boiling mesitylene or 1,2-dichlorobenzene) and (5)  $\rightarrow$  (6) (>95%; boiling xylene). We can now report that when a solution of the pyrimidinic azo-compound in 'mixed' xylenes (b.p. 138 °C) is heated under reflux for 2

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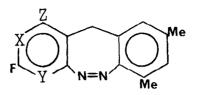
\*\* Unless stated otherwise, all yields quoted here refer to isolated materials with correct elemental compositions (% by wt.).

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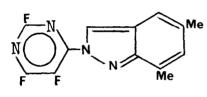
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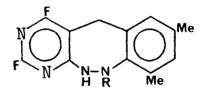


- (1) X = Y = N, Z = F(3) X = N, Y = CF, Z = F
- (5)  $X = CF, Y = N, Z = CF(CF)_{32}$



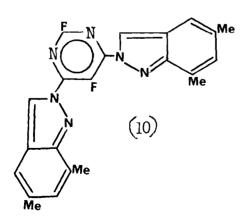
- (2) X = Y = N, Z = F
- (4) X = N, Y = CF, Z = F
- (6)  $X = CF, Y = N, Z = CF(CF_3)_2$

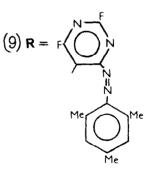




(7)

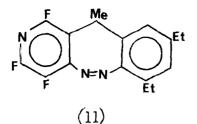


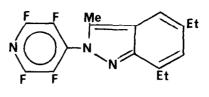




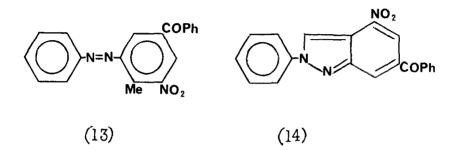
hours, almost complete (96%) conversion occurs to a mixture of 5,7-dimethyl-2-(2,5,6-trifluoropyrimidin-4-yl)-2H-indazole (7; 23%), the expected 1,2-diazepine (2; 21%), 2,4-difluoro-10,11-dihydro-7,9-dimethyl-5H-pyrimido[4,5-c]benzo[1,2]diazepine (8; 18%), and a red product (9; 4%) derived from nucleophilic attack on the starting material (1) by the last product (8). Each of these products was characterized by elemental analysis (C, H and N) and spectroscopic methods [i.r., n.m.r.  $({}^{1}H, {}^{19}F)$ , and mass]; additionally, the dihydrodiazepine (8) was oxidised to the parent diazepine (2) with mercuric oxide and with tetrabutylammonium permanganate, and cleavage of the indazole (7) with hot ethanolic sodium ethoxide was shown to give 2,4,6-triethoxy-5-fluoropyrimidine (crude yield, 51%) and 5,7-dimethylindazole (62%). Identification of the indolizinic product from the thermal reorganization of the azo-compound (1) as the 2H- rather than the 1H-isomer rests on the precedent established by Elguero's group that phenylation of 7-methylindazole with 2,4-dinitrofluorobenzene occurs only at the 2- position, a result ascribed to steric hindrance [3]: treatment of tetrafluoropyrimidine with 5,7-dimethylindazole in boiling THF gave a compound identical with that formed from (1), together with a compound logically assigned the bis-indazole structure (10).

Discussion of the mechanism of the conversion  $(1) \rightarrow (7)$  is deferred to a full presentation of the above work. Note, however, (i) that the pyrimidinylindazole (7) is stable in boiling xylene for a period (2 hours) equal to that employed in the original thermal reorganisation of azo-compound (1), i.e. the conversion (7)  $\rightarrow$  (2) appears to be ruled out; (ii) no indolizinic products have been encountered in the several other reactions where production of 1,2-diazepines hinges on the presence of an ortho-methyl substituent in the starting azo-compound [1,2]; (iii) a mixture of a pyridobenzo-1,2-diazepine (11; 23%) and a pyridylindazole [assumed to be the 2H-isomer (12) (63%)] is obtained when the triethylated azo-compound 4-(2,4,6-Et<sub>3</sub>C<sub>6</sub>H<sub>2</sub>N=N)C<sub>5</sub>F<sub>4</sub>N is heated in boiling 1,2-dichlorobenzene [2], a result which contrasts markedly with the 'clean' conversion (3)  $\rightarrow$  (4) [1] referred to earlier; (iv) thermal formation of indazoles from unsymmetrical azo-arenes containing an ortho-methyl substituent in one ring but no recognizable nucleofuge in the other has been noted previously, <u>e.g.</u> the conversion (13)  $\rightarrow$  (14) (93%) proceeds smoothly in boiling ethanol containing sodium carbonate and nitrosobenzene [4]; and (v) 5,7-dimethyl-2-(2,5,6-trifluoropyrimidin-4-yl)-2<u>H</u>-indazole (7) is formed, together with several as yet unidentified products but none of the possible pyrimidino-diazepine (2), when a solution of the





(12)



(phenylazo)pyrimidine (1) in toluene is boiled (111 °C) for 8 hours or kept for a long period in the dark at room temperature (7 is easily detected by  $^{19}$ F n.m.r. after 91 days).

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- A.C. Alty, R.E. Banks, B.R. Fishwick, R.G. Pritchard and A.R. Thompson, <u>J.Chem.Soc., Chem.Commun</u>., (1984) 832.
- 2 A.C.Alty, R.E. Banks, B.R. Fishwick and A.R. Thompson, <u>Tetrahedron Lett.</u>, 26 (1985) 1345.
- J.Elguero, A.Fruchier and R. Jacquier, <u>Bull.Soc.chim. France</u>, (1967) 2619.
- L. Chardonnens and P. Heinrich, <u>Helv.Chim.Act</u>., 23 (1940) 1399; L.
  Chardonnens and M. Buchs, 29 (1946) 872.